Influence of Remineralizing Agents on the Surface Roughness of Eroded Dental Enamel: in Vitro Study

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Abstract

The objective of this in vitro study was to evaluate the influence of remineralizing agents on the surface roughness of eroded tooth enamel. Forty incisive bovines were selected and sectioned to obtain specimens of 4mm X 4mm, resulting in a surface area of 16.00mm², with a thickness of 2.50mm. The specimens were immersed in lemon juice, twice a day, over 5 days, and randomly divided into the following 4 groups: G1 = No treatment; G2 = 1.23% APF; G3 = Fluoride Varnish 5% and G4 = Regenerated. The surface roughness of the specimens was analyzed by means of confocal laser scanning microscopy. The obtained data were submitted to ANOVA after attend the homogeneity (Levene’s) and normality tests (Kolmogorov-Smirnov). All tests were statistically significant at 5% (α=0.05). The results showed that there was no statistically significant difference on the enamel roughness surface area between groups in the experimental region and the founded values were higher at the experimental when compared to the control region, however, without differences between the groups in each region. It was concluded that the remineralizing agents were not able to reduce the surface roughness of the studied groups.

Keywords: Tooth Erosion. Dental Enamel. Fluorine. Tooth Remineralization.

1 Introduction

Dental erosion is defined as the irreversible loss of tooth structure due to the effect of acidic substances on the teeth surface without bacterial involvement.1 Initiated at the enamel, the erosion can advance into dentin because the acids dissolves the inorganic dental hard tissue what cause the increase of the surface roughness resulting in the decrease of the microhardness and the mechanical resistance of the superficial dental layer.2,3

Non-bacterial acids that causes dental erosion could be from gastric acids in patients with anorexia nervosa, bulimia, hyperthyroidism and gastroesophageal disorder or others.4 Furthermore, the changes in dietary patterns with the increase of the consumption of acidic/erosive food and beverages like juices, sodas, isotonic and energy drinks affect the incidence of dental erosive disease.4

The eroded enamel presents increase in the surface roughness that increase the dental susceptibility to new acidic etching and the abrasiveness to the opposite teeth besides to facilitate the formation and adhesion of bacterial biofilm, what rises the occurrence and the progression of dental caries and periodontal diseases.5,6

The consolidated prevention treatment to dental erosion is the topical application of different fluoride compounds like dentifrice, mouth rinse solutions and gels, and varnishes, to increase the resistance of teeth to acid challenges.7 Fluoride compounds are effective in dental remineralization increasing tooth acid resistance after an acid challenge8 because in high
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concentrations they promote the precipitation of a calcium fluoride layer in the enamel\(^8\) that acts as a barrier on this surface, acting as a fluoride reservoir inhibiting contact of the acids with the dental surface.\(^8,10-13\)

Furthermore, the use of a calcium silicate and sodium phosphate-based toothpaste presented great results as a treatment to erosive dental lesions because its regular use can promote mineral regeneration at the enamel surface.\(^8\)

Thus, the present study analyzed the influence of different remineralizing agents on the treatment of acid erosion in dental enamel by the evaluation of surface roughness (µm\(^2\)) with confocal laser scanning microscopy. The null hypothesis of this study was that the erosive challenge and the different surfaces treatments would not result in statistically significant differences in surface roughness values.

2 Material and Methods

2.1 Teeth Selection

Forty specimens were obtained from forty bovine incisors that were selected without the presence of cracks and wear. The teeth were cleaned and then immersed in a 10% formalin solution (pH = 7) for 7 days for disinfection. These teeth were then washed and stored in distilled and deionized water at a temperature of 4ºC, changed daily for a period of 7 days.

2.2 Preparation of Specimens

The incisors were selected by separating the coronary portion of the root using a refrigerated diamond disk on a cutting machine. The first cut was made 1 mm above the enamel-cement junction. The second cut was performed in the mesiodistal, obtaining three parts (cervical, middle and incisal third). The middle third of the crown was resected to obtain specimens in the initial dimensions of 4.25mm x 4.25mm. The specimens had their sides adjusted using # 600 sandpaper under cooling water to 4mm x 4mm standardization, resulting in a surface area of 16mm\(^2\). Exterior surface polishing (vestibular) was performed on polishing machine using the following sanding sequence: # 320 for 15 seconds; # 600 for 20 seconds, # 1200 for 30 seconds. Finally, felt and alumina disc polishing was performed for 30 seconds for the initial standardization of all specimens. Variations in size were allowed by 10%, plus or minus.

Half of the surface of each specimen was covered with adhesive tape. Two layers of red cosmetic nail polish enamel and sculpt wax were applied, making the waterproofing of the specimens. After this procedure, the insulating tape was removed, and each specimen was left with half of the free surface of the protection made with nail polish enamel and wax.

The specimens were stored in distilled and deionized water at a temperature of 4 ºC and the erosive challenge was performed on the specimens that were randomly divided into 4 groups (n = 10), where each group received their treatment.

2.3 Erosive Challenge

The specimens were submitted to erosive challenge in lemon juice (Del Valle - The Coca-Cola Company, Ribeirão Preto, SP, Brazil) that presented the pH of 2.8 at the temperature of 10 ºC (PG1800 digital pHmeter, GEHAKA). Each group was placed separately on a Becker for 1 minute on a magnetic stirrer. After, the erosive solution was discarded, and the specimens washed with distilled and deionized water for 10 seconds and stored again in distilled water and placed in the oven at 37 ºC between cycles. This procedure was performed twice a day, with minimum intervals of 2 hours between challenges, for a total period of 5 days, after that the specimens were conditionate at a distilled water until the next day, when the different proposed treatments were performed.

2.4 Specimen treatment

The treatments were (n=10): G1- no treatment (control group); G2- 1.23% fluoridated gel; G3- 5% fluoride varnish; G4- REGENERATE®.

The 1.23% Acidulated Phosphate Fluoride (APF) (Fluoride Gel - Nova DFL) and 5% Sodium Fluoride Varnish, Duraphat® were applied at the surface of its respective samples with a disposable applicator (Microbrush®) for 4 minutes, and the REGENERATE Enamel Science™ Biphasic Silicate/Phosphate Gel were applied for 1 minute at the samples. The excesses of the products were removed with a sterile gauze.

2.5 Surface roughness analysis

After the erosive challenge, followed by the application of the surface treatments agents, the topographic profile 3D was analyzed (wear profile), with the purpose of checking the surface roughness in the different experimental groups by a confocal laser scanning (OLS4000®).

2.6 Statistical Analysis

The obtained data were analyzed for distribution (Levene) and normality (Kolmogorov-Smirnov). Once these conditions were satisfied, the statistical analysis of variance (ANOVA) was performed, with a significance level of 5% (α = 0.05) using the SPSS 17.0 software.

3 Results and Discussion

The surface roughness results (µm\(^2\)) of the present study are described in Table 1 below.

<p>| Table 1 - Mean values (standard deviation) of the surface roughness of area (µm(^2)) of the control and experimental (after treatments) specimen regions for the groups |
|---------------------------------|-------------------|-------------------|</p>
<table>
<thead>
<tr>
<th>Groups</th>
<th>Control Region</th>
<th>Experimental Region</th>
</tr>
</thead>
<tbody>
<tr>
<td>G1</td>
<td>0.451 (0.032)(^a)</td>
<td>2.587 (0.087)(^b)</td>
</tr>
<tr>
<td>G2</td>
<td>0.412 (0.027)(^a)</td>
<td>2.612 (0.054)(^b)</td>
</tr>
<tr>
<td>G3</td>
<td>0.433 (0.039)(^a)</td>
<td>2.512 (0.094)(^b)</td>
</tr>
<tr>
<td>G4</td>
<td>0.457 (0.035)(^a)</td>
<td>2.497 (0.107)(^b)</td>
</tr>
</tbody>
</table>

\(^a\) different letters represent statistically significant difference (p <0.05). 

Source: Research data.
In the control region, it was observed that there was no statistically significant difference between the groups. For the experimental region, there was an increase in surface roughness values for all groups with a statistically significant difference when compared to the control region (p <0.05), but without differences between them.

After the surface roughness analysis results, it was found that the null hypothesis of the present study was partially rejected, since the roughness values found after erosion were higher when compared to the control region and the surface treatments were not effective to recover the enamel integrity of the specimens.

In the present study, bovine teeth were used at the experimental test as a substitute for human teeth because the chemical composition and physical features of bovine enamel are similar to those of human teeth. In addition, bovine incisors are frequently used in dental research due to the difficulty to obtain intact human teeth, and its larger surface allow the use of demineralization and remineralization researches.

In this study it was observed that there was no significant difference in statistical values between the groups of the control region, that indicates that the waterproofing of the specimens with the nail enamel and sticky wax was effective (Table 1). However, in the experimental region of the specimens the values of surface roughness presented statistical similar values between the groups and statistically higher when compared with that obtained at the control region (Table 1).

The founded results show that the use of the lemon juice as a demineralizing agent attempting to cause extrinsic factor erosion in tooth enamel was effective due to exposure to acid juice since acid beverage consumption is one of a causal factors of dental erosion. In the present study the pH of the erosive drink used was 2.8 at a temperature of 10°C, below to 5.5 and 6.5 that are the critic pH of enamel and dentin, respectively, thus showing that the utilized juice has an acidic pH enough to cause tooth demineralization by increasing the surface roughness at the enamel. It probably occurs because at low pH like that observed for the lemon juice, the amount of H+ increases, wakening the binding of Ca+2 ions, leading then into solution, changing the enamel surface by the increasing at the surface roughness, what was observed in this study (Table 1). Other authors found by SEM images analyses that citric fruit juice caused morphological changes to the enamel surface, including the presence of craters and surface irregularities.

The tested treatments for the erosion in this study were the application of 1.23% APF, 4% fluoride varnish and Regenerate, which has been used as a clinical treatment of demineralized enamel. However, the tested materials were not capable to repair the surface roughness of the specimens after the acid challenge. This can be explained because the fluoride treatments can form fluoridated apatites in the enamel structure and accelerate the deposition of CaF2-like material and calcium phosphate from the supersaturated gel.

Besides, the fluoride can result in lower pH, making calcium available for interaction with fluoride in the form of a layer of CaF2. Conversely, although the calcium silicate and sodium phosphate-based material can to release calcium ions to the surrounding oral fluids under acid conditions, to inhibit the enamel hydroxyapatite solution and to buffer the protons absorption at the eroded enamel surface, in the present study, the Regenerate application did not promote the decrease of the surface roughness at the enamel. This can be justified because of the protocol utilized in this study, where it was done one application of the material, being indicate the continuous use to obtain the recovering of the integrity of the enamel damage surface.

The preventive effects of different types of fluoride on erosion injuries have positive results. As the unsuccessful obtained results in the present study to fluoride and calcium silicate and sodium phosphate-based materials, an alternative treatment could be adopted is the enamel erosion prevention. The analyzed materials did not have the ability to naturally restore the tooth with erosion already in place, however, they could be used to prevent the enamel erosion.

4 Conclusion

Considering the results obtained in the present study, remineralizing agents were not able to reduce the surface roughness of the studied groups, demonstrating the importance of adopting preventive strategies.

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